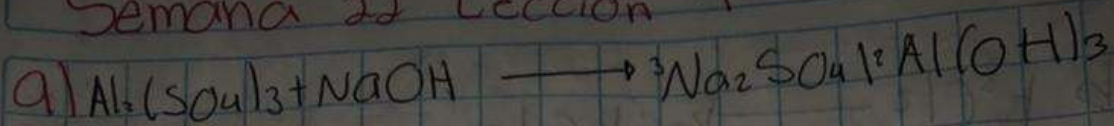


Semana 22 Lección 4



$$\text{Al} = 2$$

$$\text{Na} = 1 = 6$$

$$\text{S} = 3$$

$$\text{H} = 1 = 6$$

$$\text{O} = 6 + 12 = 18$$

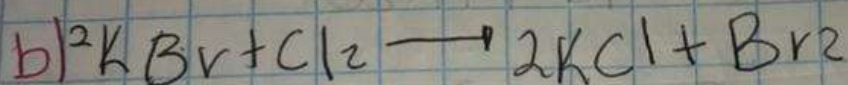
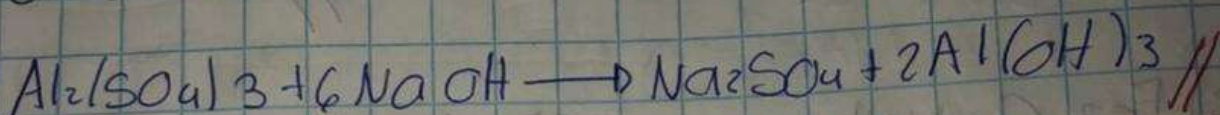
$$\text{Al} = 1 = 2$$

$$\text{Na} = 2 = 6$$

$$\text{S} = 1 = 3$$

$$\text{H} = 3 = 6$$

$$\text{O} = 6 + 12 = 18$$



$$\text{K} = 1 = 2$$

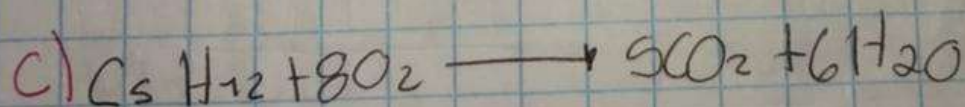
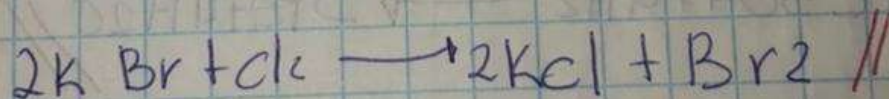
$$\text{Br} = 1 = 2$$

$$\text{Cl} = 2$$

$$\text{K} = 1 = 2$$

$$\text{Br} = 2$$

$$\text{Cl} = 1 = 2$$



$$\text{C} = 5$$

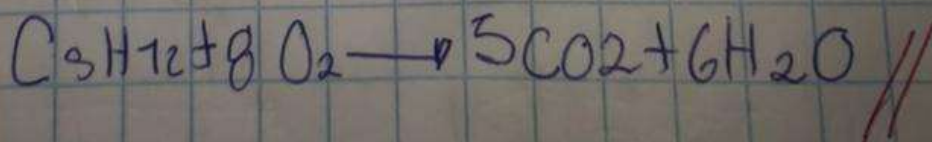
$$\text{H} = 12$$

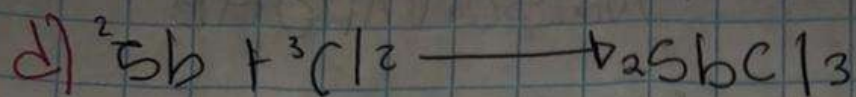
$$\text{O} = 2 = 16$$

$$\text{C} = 1 = 5$$

$$\text{H} = 2 = 12$$

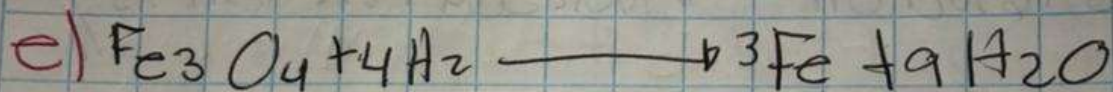
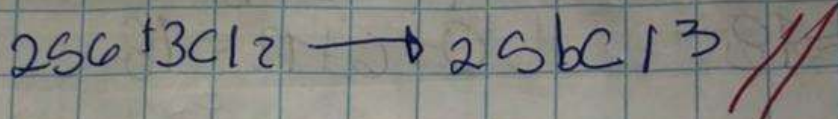
$$\text{O} = 10 + 6 = 16$$





$$\begin{aligned}\text{Sb} &= 1 = 2 \\ \text{Cl} &= 2 = 6\end{aligned}$$

$$\begin{aligned}\text{Sb} &= 1 = 2 \\ \text{Cl} &= 3 = 6\end{aligned}$$



$$\begin{aligned}\text{Fe} &= 3 \\ \text{H} &= 2 = 8 \\ \text{O} &= 4\end{aligned}$$

$$\begin{aligned}\text{Fe} &= 1 = 3 \\ \text{H} &= 2 = 8 \\ \text{O} &= 1 = 4\end{aligned}$$

