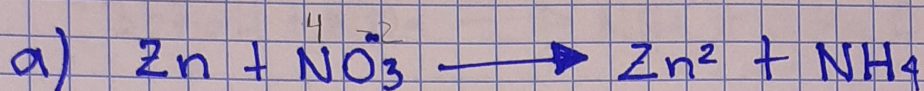
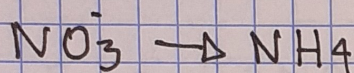
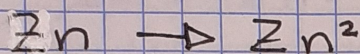


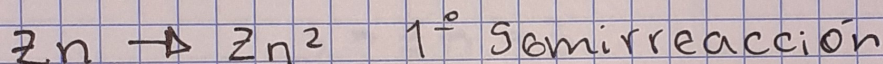
Balancear las siguientes ecuaciones redox solución ácido



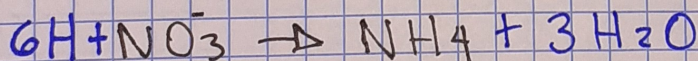
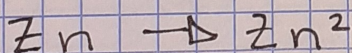
• Paso # 1



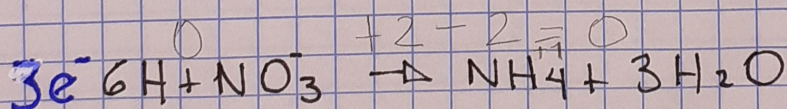
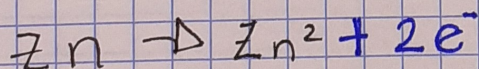
• Paso # 2



• Paso # 3



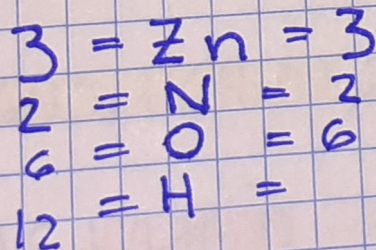
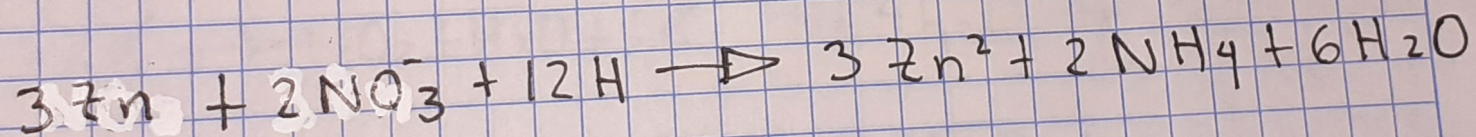
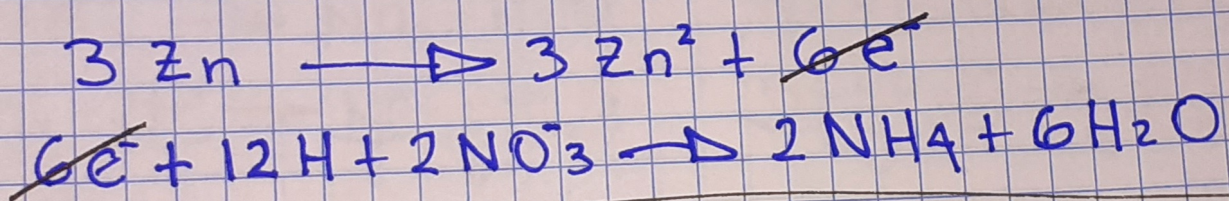
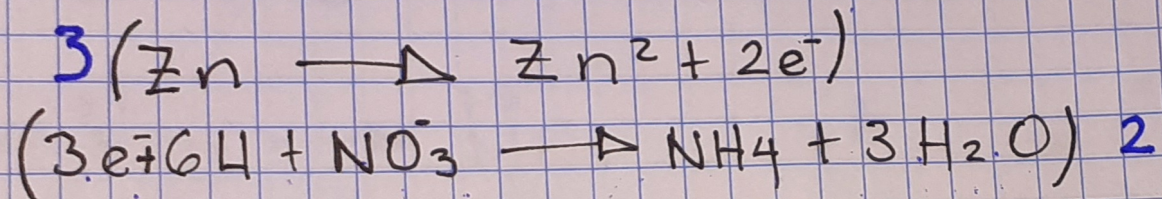
• Paso # 4



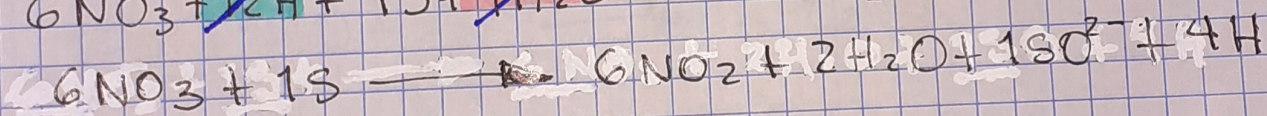
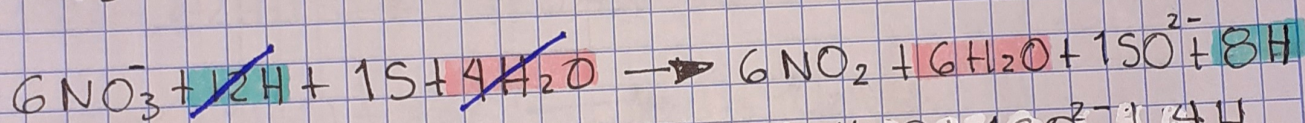
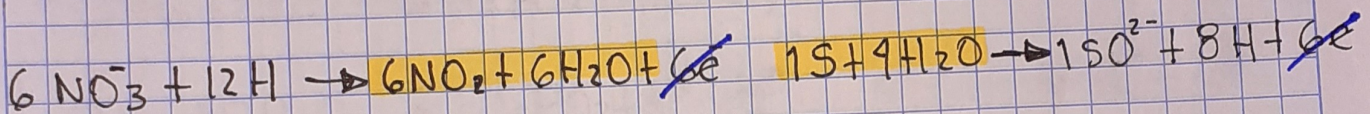
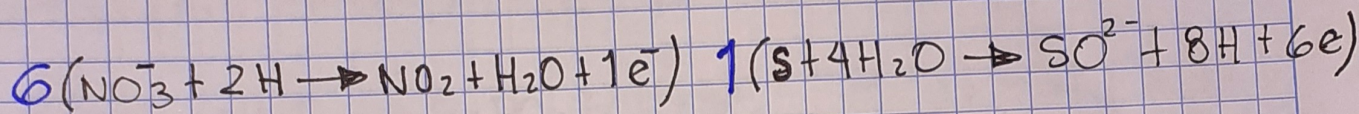
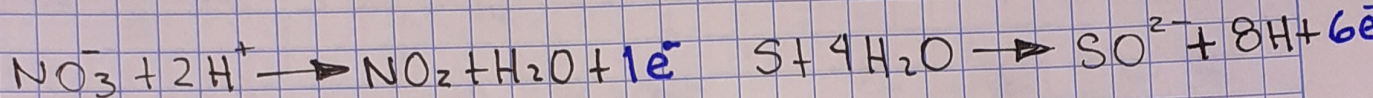
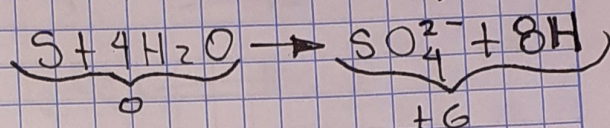
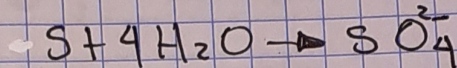
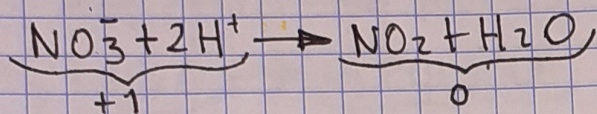
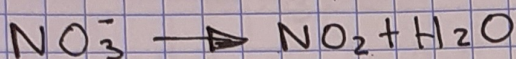
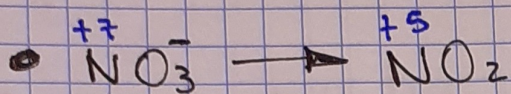
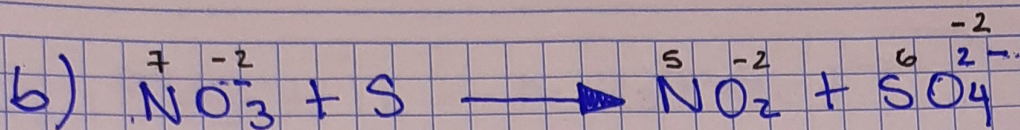
$$-3 + 6 - 1 = 5 \quad (1 + 0 = +1)$$



Paso # 5







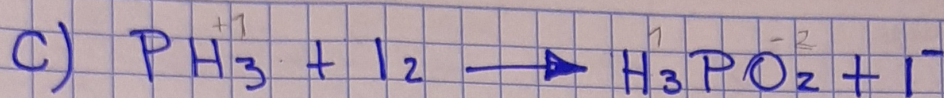
$$6 = \text{N} = 6$$

$$1 = \text{S} = 1$$

$$= \text{O} =$$

$$= \text{H} = 8$$

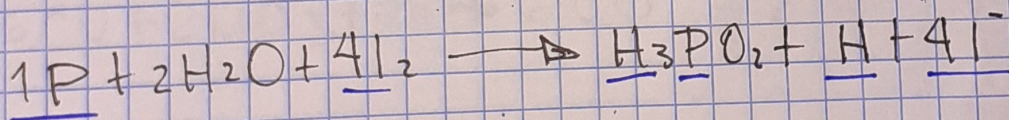
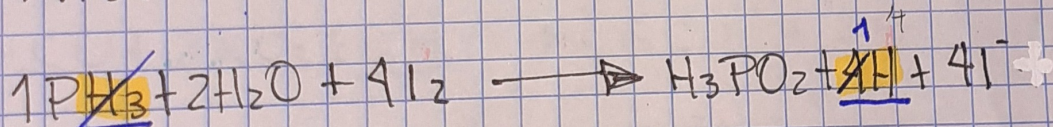
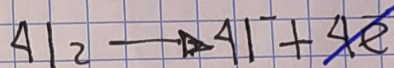
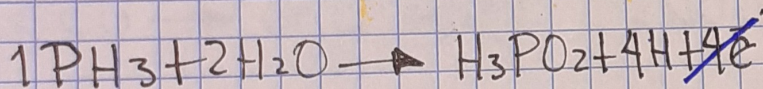
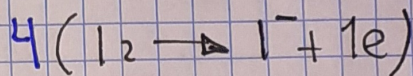
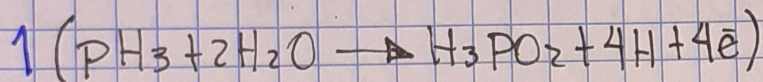
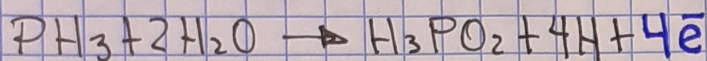
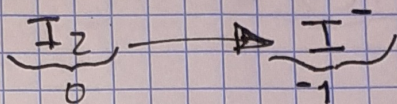
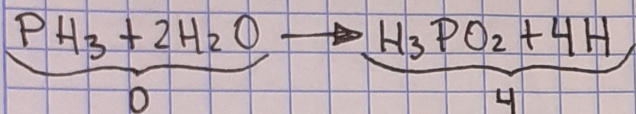
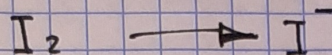
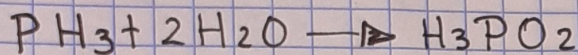
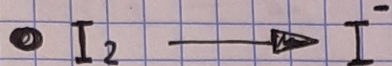
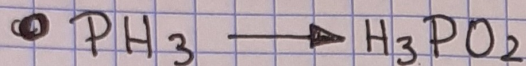




# 1

3

-4



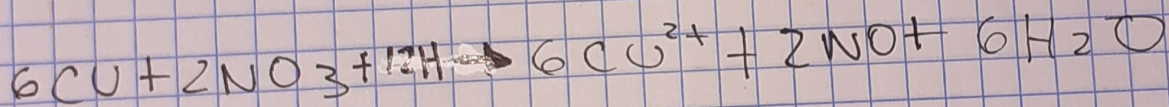
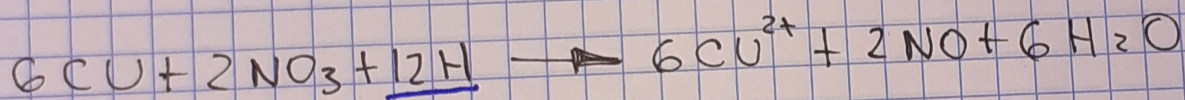
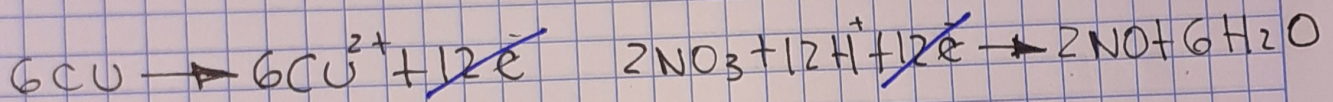
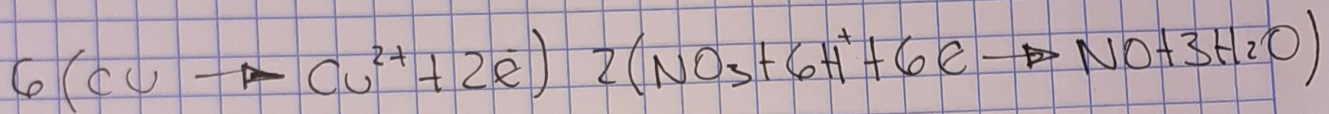
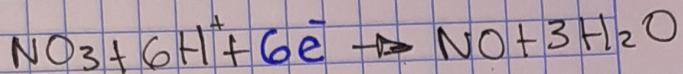
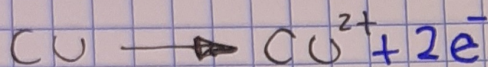
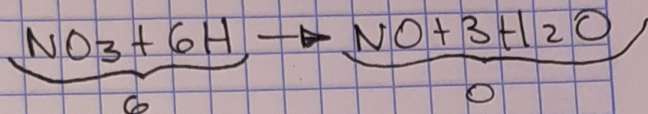
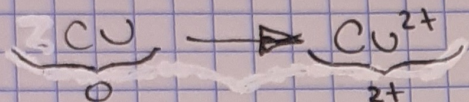
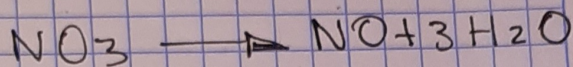
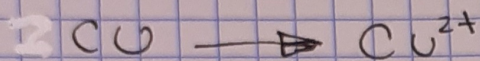
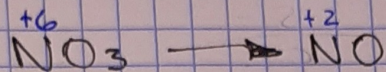
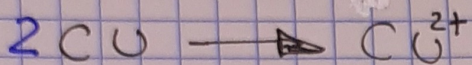
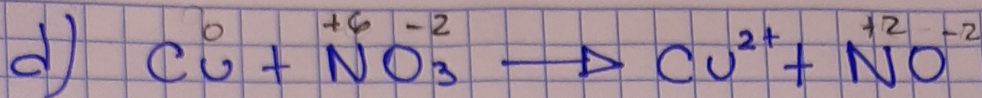
$$1 = P = 1$$

$$4 = I = 4$$

$$4 = H = 4$$

$$2 = O = 2$$





$$6 = \text{Cu} = 6$$

$$2 = \text{N} = 2$$

$$6 = \text{O} = 6$$

$$12 = \text{H} = 12$$